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$3\text{Cu} + 8\text{HNO}_3 \rightarrow 3\text{Cu}(\text{NO}_3)_2 + 2\text{NO} + 4\text{H}_2\text{O}$ It is a colourless gas, insoluble in water, which reacts with oxygen to form the brown gas nitrogen dioxide. $2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$. Nitrogen dioxide: Nitrogen dioxide, NO_2 is a planar molecule. $3\text{Cu}(\text{NO}_3)_2 + 2\text{H}_2\text{O} + 2\text{NO}_2$ or by the decomposition of heavy-metal nitrates such as lead nitrate: $2\text{Pb}(\text{NO}_3)_2 \rightarrow 2\text{PbO} + 4\text{NO}_2 + \text{O}_2$ At temperatures below ...

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